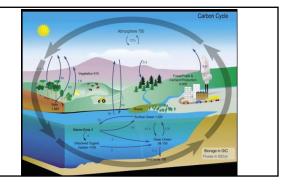
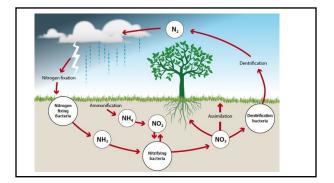
Atmospheric Pollution

APES- Unit 7

Review, Earth's Cycles





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What are Air Pollutants? Definition-is the presence of Examples substances in the atmosphere that Carbon Dioxide are harmful to the health of humans Sulfur Dioxide and other living beings, or cause Coal Combustion damage to the climate or to Toxic Metals materials. Particulates > Nitrogen Oxides Other Fossil Fuels 3 types- Gases, Biological Carbon Monoxide Molecules, Particulate Matter (PM) Hydrocarbons





What does Air Pollution Cause?

- Can Lead to the production of Ozone
- Can lead to depletion of Ozone
- Formation of Photochemical smogAcid Rain
- Air Quality Issues
- Human Health Problems
- Animal Health Problems
- Soil Health ProblemsGlobal Warming



Primary Vs. Secondary Pollutants

Primary Pollutants

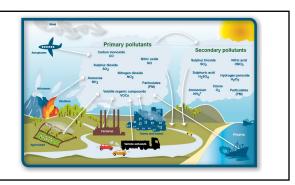
Released directly into the lower atmosphere and are toxic

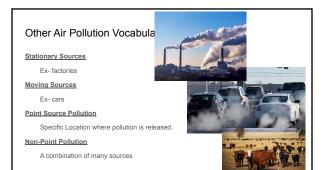
Ex-Carbon Monoxide

Secondary Pollutants

Those that are formed by the combination of primary pollutants in the atmosphere

Ex- Acid rain





Who is protecting the Air?

Environmental Protection Agency (EPA)

National Ambient Air Quality Standards (NAAQS)

- Clean Air Act
- Montreal Protocol (1986)

• Energy Policy & Conversion Act (1975)

Ozone

What is it? Is it a good, or bad?

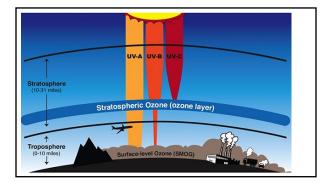


O₂

Stratospheric Ozone - Absorbs UV light from the sun. Protects life on

<u>Tropospheric Ozone-</u> A powerful respiratory irritant & precursor to secondary air pollutants

*Up High Ozone helps us; down low, it hurts us



Chlorofluorocarbons (CFCs)

*Invented in the 1930s, CFCs and other related compounds are responsible for Ozone Depletion in the Stratosphere.

 $^{\star}\text{CFCs}$ were used in propellants, fire extinguishers, and cans of Hairspray.

*Once they reach the stratosphere, UV light breaks the molecules apart to form chlorine atoms. Creating Chlorine Monoxide

$$CI + O_3 \rightarrow CIO + O_2$$

Photochemical Smog

What is Photochemical Smog?

- "Brown Smog"
- Hot, sunny days- the intensity of the sun matters
- Urban Areas
- NO_x Compounds, VOCs, & Ozone all combine
- Las Angeles, California and Athen,
 Greece are the most susceptible cities



What contributes to Photochemical Smog?

 ${\rm 2NO+} \; {\rm O_2} \! \to {\rm 2NO_2} \; \; ({\rm Causes \; brownish \; Haze})$

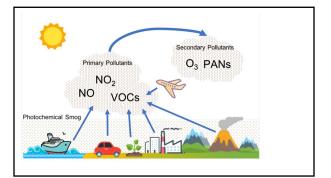


 $NO_2 + UV Light \rightarrow NO + O$



 $O+O_2 \rightarrow O_3$

 $\label{eq:hydrocarbons+O2+NO2} \operatorname{Hydrocarbons+O2+NO_2} \operatorname{->PANS} \ (\operatorname{Peroxyacyl} \ \operatorname{Nitrates})$



What is the long term health concerns of Photochemical

Smog?
Ozone- is hazardous to plants, animals, and materials in the troposphere

Peroxyacyl Nitrates-causes burning eyes & damage to vegetation



Examples of Photochemical Smog Events

The great Smog 1952- London England, December 5th-9th, nearly 12,000 Deaths

2013 Southeast Asian Haze- Singapore reached a hazardous range of 401

Thermal Inversion

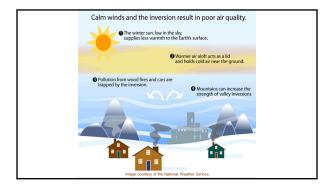
What is Thermal Inversion?

In this phenomenon, air pollutants become

This happen over any city where a large mass of warm air can become stalled



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Atmospheric ${\rm CO_2}$ & Particulates

The Natural sources of ${\rm CO_2}$ & Air Particulates



CO₂ sources

- Respiration
- Decomposition
- Volcanic Eruptions
- Mother Nature caused Forest Fires



Air Particulate Sources

- Sea Salt
- Dust (Airborn Soil)
- Pollen
- Volcanic Eruptions
- Natural Gaseous Precurors



What is the oposite of a natural source?

Anthropogenic

What is the scientific name for Man-Made?

*Anthropo-, a prefix meaning human, humanoid, human-like.

Indoor Air Pollutants

What are indoor air pollutants?

Indoor Air Quality (IAQ) refers to the air quality within and around buildings and structures, especially as it relates to the health and comfort of building occupants.



- Asbestos Dust Smoke Radon Mold Carbon Monoxide

- Nitrogen Oxides
 Suffur dioxide
 Tobacco Smoke
 Volatile Organic Compounds (VOCs)

Wait..what's radon and asbestos?

Radon-222

Naturally occurring radioactive gas, produced by the decay of uranium found in some rocks and





Naturally occuring mineral composed of soft and flexiable fibers that are resistant to hear, electricty and corrosion.

Whore	40	indoor	Λir	Dollutanto	come from	2
vvnere	uυ	macon	ΑII	Pollulanis	come nom	•

EVERYTHING

Natural Source

Human Made Sources

Combustion

- Insulation
- Furniture
- Carpets
- Paints
- Upholstery
- Cigarettes
- Vape
- Heaters

- Soil
- Rocks

- Paneling

- Stoves
- Foundation
- Wells

What's the effect of indoor air pollutants?

Shorter Term Effects

- Asthma & other respitory problems
 Headaches, Dizzyness, memory loss,
- lethargy
 Frequent Colds & Sore thoats
 Skin Rashes
- Skin RasnesEye Irriation

Long Term Effects

- Exposure to Radon gas can lead to radon
- induced lung cancer

 Exposure to Asbestos can cause cancers
- and lung deaseses

 Lung Cancer from Smoke Inhilation

 Chronic Bronchitis

Sick Building Syndrome

What is SBS (sick building syndrome)?

SBS is the term used when the majority of a building's occupants experience certain symptoms.

Hard to Diagnose



Preventing SBS Making sure buildings are properly ventilated Limiting exposure to certain chemicals & Cleaning Fluids Don't smoke indoors WARTE GUALITY WARTE GUALITY

Acid Rain

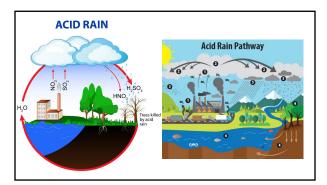
What is Acid Rain?

Its real name is Acid Deposition









Show me the Chemistry! Sulfur Oxides

Sulfur dioxide (SO₂) is produced industrially from the combustion of sulfur-containing fossil fuels and smelting of sulfide ores

 $\mathrm{S}(s)+\mathrm{O}_2(g)
ightarrow \mathrm{SO}_2(g)$

2) Sulfur dioxide (SO2)is then oxidized by sunlight to form sulfur trioxide $\,$

 $2\mathrm{SO}_2(g) + \mathrm{O}_2(g) o 2\mathrm{SO}_3(g)$

(SO₃)

 $\mathrm{SO}_2(g) + \mathrm{H}_2\mathrm{O}(l) o \mathrm{H}_2\mathrm{SO}_3(aq)$

3)The oxides react with water to form

 $\mathrm{SO}_3(g) + \mathrm{H}_2\mathrm{O}(l) o \mathrm{H}_2\mathrm{SO}_4(aq)$

Show me the Chemistry! Nitrogen Oxides

1) Nitrogen monoxide (NO) is produced from internal combustion of engines

 $N_2(g) + O_2(g) \rightarrow 2NO(g)$

2) Nitrogen monoxide (NO) oxidizes to form nitrogen dioxide

 $2\mathrm{NO}(g) + \mathrm{O}_2(g) \to 2\mathrm{NO}_2(g)$

3) Nitrogen dioxide reacts with vater to form nitric acid (HNO3)and nitrous acid (HNO2) or it reacts with oxygen and water and becomes nitric acid

 $2\mathrm{NO}_2(g) + \mathrm{H}_2\mathrm{O}(l) \to \mathrm{HNO}_3(aq) + \mathrm{HNO}_2(aq)$

 $4\mathrm{NO2}_g + \mathrm{O_2}(g) + \mathrm{H_2O}(l) \rightarrow 4\mathrm{HNO_3}(aq)$

The impact of Acid Deposition

- Mainly affects communities that are down wind from coal-burning power plants
- Can lead to acidification of soils and Bodies of water
- Corrosion of human-made structures
- Regional differences in soil and bedrock affect the impact
 - Limestone bedrock had the ability to neutralize the effect of acid rain on lakes and ponds.
- Creating a buildup of sulfur and nitrogen
- Leaching calcium ions from the needles in
- conifers
- Elevating the aluminium concentration in lakes, that are toxic to fish
- Lowering the pH of streams, rivers, lakes, may lead to fish death
- Causing human respiratory irritation

Examples of Acid Rain Noise Pollution What is Noise Pollution? **Noise Pollution** is sound at levels high enough to cause physiological stress and hearing loss

Sources of Noise Pollution in Urban Area's

Transportation
Construction
Domestic and Industrial activity

Sources in Marine Ecosystems

Boats
Sonar
Old & Gas Drilling

The Impact of Noise Pollution on	
Animal Ecosystems Stress Masking of sounds used to Cause anxiety communicate or hunt Poor concentration	
Damaged hearing Causing Changes to Migratory Routes Stress Loss of productivity Lack of sleep Stress	
Deduction of Air Dellutonte	
Reduction of Air Pollutants	
The Three Methods to Reduce Air Pollutants	
Regulation Practices 2. Conservations Practice	
3. Alternative Fuels	

Wet & Dry Scrubbers

Industrial Scrubbers are devices that can control pollution by puriying the exhust streams

*Both types of industrial scrubbers are able to eliminate acidic gases that are known to directly contribute to harmful acid rain Wet- remove harmful substances from exhaust gases by using a west substance to do so. When flue gas is sent through the system, the wel industrial scrubber sprays this gas with a special fluid that reacts mainly with the heavier contaminants that are present in the gas

<u>Dry-</u> sprays numerous dry reagents onto the exhaust stream. This slurry can either neutralize pollutants or change them into a different substance altogether

Catalytic Converter-

Converts pollutants (CO, NOx, & Hydrocarbons) into less harmful molecules (CO₂, N₂, O₂ and H₂O)

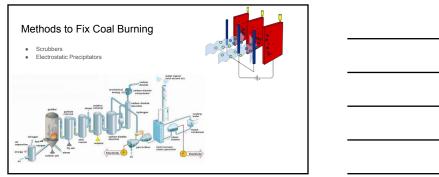


Vapor Recovery Nozzle



They capture the gasoline vapors that escape from automobile tanks when they are being refueled

The vapors are returned to the underground gasoline storage tank through special hoses and pipes.



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